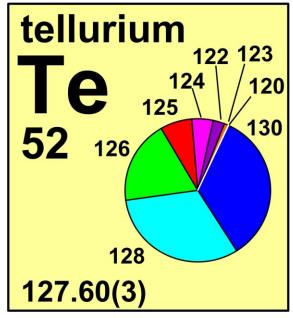
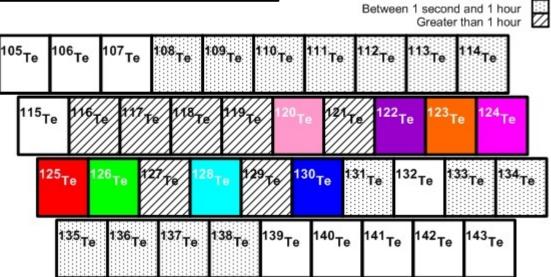
## tellurium



Stable	Atomic mass*	Mole
isotope		fraction
<sup>120</sup> Te	119.904 02	0.0009
<sup>122</sup> Te	121.903 0439	0.0255
<sup>123</sup> Te	122.904 27	0.0089
<sup>124</sup> Te	123.902 8179	0.0474
<sup>125</sup> Te	124.904 4307	0.0707
<sup>126</sup> Te	125.903 3117	0.1884
<sup>128</sup> Te	127.904 4631	0.3174
<sup>130</sup> Te	129.906 2244	0.3408

<sup>\*</sup> Atomic mass given in unified atomic mass units, u.

Half-life of redioactive isotope Less than 1 second



## Important applications of stable and/or radioactive isotopes

Isotopes in medicine

- 1) <sup>120</sup>Te is used for the production of <sup>120g</sup>I which has an application as a PET and Beta emitting isotope.
- 2) 122 Te is used in the production of the radioisotope 122 I which is used in gamma imaging.
  3) 123 Te is used for the production of radioactive 123 I which is used in thyroid imaging.
  4) 124 Te is used for the production of both 123 I and the PET isotope 124 I.

- 5) <sup>130</sup>Te is used in the research into double Beta decay.

<sup>\*\*</sup>Applications of tellurium isotopes are still being researched and this page will be updated shortly. \*\*